

## Chemistry Worksheet Stoichiometry Mixed Problems 5 Answers

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### Chemistry Worksheet Stoichiometry Mixed Problems

Stoichiometry: Mixed Problems (KEY) 1)  $N_2 + 3H_2 \rightarrow 2NH_3$  What volume of  $NH_3$  at STP is produced if 25.0 of  $N_2$  is reacted with an excess of  $H_2$ ? 33.3 L  $NH_3$  2)  $40.0L NH_3$  1mol  $NH_3$  22.4L  $NH_3$  1mol  $N_2$  2mol  $NH_3$  28.0g  $N_2$  25.0g  $N_2$  1mol  $N_2$   $\times \times \times =$  2)  $2KClO_3 \rightarrow 2KCl + 3O_2$  If 5.0g of  $KClO_3$  is decomposed, what volume of  $O_2$  is produced at STP? 2

### Stoichiometry: Mixed Problems (KEY)

4. How many mL of 0.5 M CsBr are required to produce 100 g of CsOH? (Use the balanced equation in problem 3). 1340 mL Mixed

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Stoichiometry Problems 1. How many moles of H<sub>2</sub> would be required to completely react with O<sub>2</sub> to produce 5 moles of water? 5 mol H<sub>2</sub>

2. H<sub>2</sub>SO<sub>4</sub> + NaOH → Na<sub>2</sub>SO<sub>4</sub> + H<sub>2</sub>O a. Balance this equation b.

## Stoichiometry mixed Problems 1011

Name: \_\_\_\_\_ Stoichiometry - Mixed Problems 1. N<sub>2</sub> + 3H<sub>2</sub> → 2NH<sub>3</sub> What volume of NH<sub>3</sub> at STP is produced if 25.0g of N<sub>2</sub> is reacted with an excess of H<sub>2</sub>? 2. 2KClO<sub>3</sub> → 2KCl + 3O<sub>2</sub> If 5.0 g of KClO<sub>3</sub> is decomposed, what volume of O<sub>2</sub> is produced? 3. How many grams of KCl are produced in Problem 2?

## Stoichiometry - Mixed Problems - Mr. V's Chemistry Site

Stoichiometry - Mixed Problems. This is the last of the series of four stoichiometry worksheets. This one mixes several different types of problems -- moles to moles, moles to grams, grams to grams, and even some conversions with particles and volume.

## Stoichiometry Mixed Problems Worksheet |

### Aurumscience.com.

Title: Microsoft Word - 8-14a,b Mixed Problems--Mole-Mole and Mole-Mass wkst-Key.doc Author: Brent White Created Date: 7/13/2005 9:53:18 PM

**2 mol C H ? mol C H = 5.5 mol O = 0.85 mol C H 13 mol O**

...

Chemistry: Stoichiometry - Problem Sheet 2 KEY 9) 2 24 2 2 23 2 2 2 4.63 x 10 molecules | 1 mol | 6.02 x 10 molecules | 1 mol Cl 1mol 71 g Cl Cl x 546 g Cl 10) 292 g Ag 1 mol Ag 108 g Ag 1 mol Cu 1 mol Ag 63.5 g Cu 1 mol Cu x g Ag 86 g CuO 11) 3 3 3 3 2 2 2 3 2 15.7 dmNH 1 mol NH 22.4 dmNH 1 mol Ca(OH) 2mol

## Stoichiometry: Problem Sheet 2 - FREE Chemistry Materials ...

Purpose: In all of the stoichiometry problems so far, students have been given a volume, mass, or amount of one specific substance and asked to solve based on that. This worksheet gives them two measurements. They must determine which of the two is the limiting reagent -- the one that will be used up first in the reaction and will thus determine the amount of product

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## **Stoichiometry (Worksheet) - Chemistry LibreTexts**

Some of the worksheets below are Stoichiometry Worksheets with Answer Keys, definition of stoichiometry with tons of interesting examples and exercises involving with step by step solutions with several colorful illustrations and diagrams.

## **Stoichiometry Worksheets with Answer Keys - DSoftSchools**

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Quiz \*Theme/Title: Stoichiometry VI: Mixed Problems \* Description/Instructions ; This quiz will give you some more

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practice in solving the various kinds of ...

## **Stoichiometry : Stoichiometry VI: Mixed Problems Quiz**

Mixed Stoichiometry Problems . 1.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ . a). How many moles of  $\text{H}_2$  would be required to produce 5.0 moles of water? 5.0 moles water. b). What mass of  $\text{H}_2\text{O}$  is formed when  $\text{H}_2$  reacts with 384 g of  $\text{O}_2$ ? 432g  $\text{H}_2$ . 2.  $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$ . a). Balance this equation. Look above. b).

## **Mixed Stoichiometry Problems - Murrieta Valley Unified**

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Chemistry: Stoichiometry - Problem Sheet 1 Directions: Solve each of the following problems. Show your work, including proper units, to earn full credit. 1. Silver and nitric acid react according to the following balanced equation:  $3\text{Ag}(s) + 4\text{HNO}_3(aq) \rightarrow 3\text{AgNO}_3(aq) + 2\text{H}_2\text{O}(l) + \text{NO}(g)$  A.

## **Stoichiometry: Problem Sheet 1 - FREE Chemistry Materials ...**

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## **Honors Chemistry Stoichiometry Problems 1 Answers ...**

Stoichiometry Worksheet and Key  $1.65\text{ mol KClO}_3 \rightarrow 3\text{ mol KClO}_3$   
 $\text{mol O}_2 = \text{mol O}_2 \cdot 3.50\text{ mol KCl} = \text{mol KClO}_3 = 0.275\text{ mol Fe} = \text{mol Fe}_2\text{O}_3 = 2\text{ KClO}_3 \rightarrow 2\text{ KCl} + 3\text{ O}_2$  10. How ...

## **stoichiometry 1 worksheet and key - Saddleback College**

This is unlike regular solids where we only had to account for the mass of the solids and solve for the mass of the product by stoichiometry. In order to solve for the temperature, pressure, or volume of a gas using chemical reactions, we often need to have information on two out of three of these variables.

## **5.4: Gas Stoichiometry - Chemistry LibreTexts**

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### **Chemistry Mixed Stoichiometry Word Problems Answers**

Solving Stoichiometry Problems In this video, we will look at the steps to solving stoichiometry problems. 1. Start with your balanced chemical equation. 2. Convert the given mass or number of particles of a substance to the number of moles. 3.

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