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CHAPTER 21 REVIEW Nuclear Chemistry SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. ____ The nuclear equation is an example of an equation that represents (a) alpha emission. (b) beta emission. (c) positron emission. (d) electron capture. 2. ____ When undergoes electron capture to form a new element X, which of

CHAPTER 21 REVIEW Nuclear Chemistry

the conversion of atoms of one element to atoms of another. (It involves a nuclear change, not a chemical change)

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answer choices. The amount of time it takes for some of the nuclei in a sample of the isotope to decay. The amount of time it takes for half the electrons in a sample of the isotope to decay. The amount of time it takes for half the nuclei in a sample of the isotope to decay.

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Nuclear Chemistry Nuclear Transformations • Rutherford in 1919 performed the first nuclear transformation. • The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes: ${}_{14}^{22}\text{Li} + {}_2^4\text{He} \rightarrow {}_{16}^{26}\text{S} + {}_0^1\text{n}$

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Chapter 18 Nuclear Chemistry

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